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ANSWER Answer the following questions in the space provided. 1.  
Following are statements about the dissolving process. Explain each

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one at the molecular level. a. Increasing the pressure of a solute gas above a liquid solution increases the solubility of the gas in the liquid. b.

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### CHAPTER 13 AUDITING THE INVENTORY

MANAGEMENT PROCESS Answers to Review Questions 13-2

The inventory process is affected by control activities in the revenue, purchasing, and human resource management processes. The

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purchasing process controls the acquisition and payment of raw materials and overhead costs. The cost of direct and indirect labor assigned to inventory is controlled through the ...

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Chapter 13 221 Section 13.4 Solutions of Gases in Liquids Goals  
To describe the process by which gases dissolve in liquids. To explain why gases have a solubility limit in liquids. To explain why increased partial pressure of a gas over a liquid will lead to an increase in the solubility of that gas.

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following questions in the space provided. 1. Match the four compounds on the right to their descriptions on the left. b an ionic compound that is quite soluble in water (a) HCl c an ionic compound that is not very soluble in water (b)  $\text{NaNO}_3$

13 Ions in Aqueous Solutions and Colligative Properties

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CHAPTER . 12 . REVIEW "'-- Solutions . SHORT ANSWER

Answer the following questions in the space provided. 1. Describe the errors made by the following students in making molar solutions. 3. James needs a 0.600 M solution of KCl. He measures out 0.600 g of KCl and adds 1 L of water to the solid. James made several errors.

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ANSWER 1. c 2. a 3. b 2. a. alcohol b. water c. the gels 3. The  
mixture is a colloid. The properties are consistent with those  
reported in Table 3 on page 404 of the text. The particle size is  
small, but not too small, and the mixture

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33. Estimate 30 balls lost per game. 81 games  $\times$  3 1 0 g b am all e s

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= 34. Estimate 1 4 lb per ...

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